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Chapter 17 Holt chemistry vocab. **STUDY. PLAY.** Oxidation. a reaction that removes one or more electrons from a substance such that the substance's valence or oxidation state increase. Reduction. a chemical change in which electrons are gained, either by the removal of oxygen, the addition of hydrogen, or the addition of electrons.

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CHAPTER 16 REVIEW Reaction Energy SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. For the following examples, state whether the change in entropy favors the forward or reverse reaction:

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16 Reaction Energy

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160 Chapter 5 Figure 3 The periodic table shows the electron configuration of each element. The number of electrons in the outermost energy level is the number of valence electrons. Topic Link Refer to the "Periodic Table" chapter for more about valence electrons. Group 1 Group 2 Group 13 Group 14 Group 15 Group 16 Group 17 Group 18 H ...

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