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Key _____ Atomic Structure and the Periodic Table Test Study Guide Chapter 5: 1. Use the following to answer

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the questions about the fictitious element Nipomium (Nm): Color of protons: black Color of neutrons: white Reason: Isotopes must have the same number of protons…all three pictured have the same number of BLACK dots, therefore, black dots must be the protons.

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[OF MATTER](#)

Atoms are the smallest unit of matter that cannot be divided using any chemical method. They do consist of smaller parts, but can only be broken by nuclear reactions. The three parts of an atom are protons, neutrons, and electrons. Protons carry a positive electrical charge. Neutrons are electrically neutral.

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SHORT ANSWERS. 1) What is the smallest unit of matter that retains the identity of the substance? 2) Atoms are composed of two regions, what are they and what do they contain? 3) What does the Atomic Number tell you about an atom? 4) Which two particles add up to equal the mass number? 5) Solve this equation: #of neutrons=_____ # - atomic #

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structure is best based on the octet rule and formal charges
Formal Charge = $(\# \text{ of valence } e^-) - \frac{1}{2} (\# \text{ of bonding } e^-) - (\# \text{ of nonbonding } e^-)$
The more formal charges there are on the atoms in your structure, the less the

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34 Study Guide for An Introduction to Chemistry Exercise 3.3 - Cations and Anions: Identify each of the following as a cation or an anion, and determine the charge on

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each. (Obj 22) a. magnesium atom with 12 protons and 10 electrons $(+12) + (10) = +2$ This is a +2 cation. b. fluorine atom with 9 protons and 10 electrons

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Unit 2 Atomic Structure & Periodic Table Study Guide Answer Section MULTIPLE CHOICE 1. ANS: A PTS: 1 2. ANS: B PTS: 1 3. ANS: D PTS: 1 4. ANS: C PTS: 1 5. ANS: C PTS: 1 6. ANS: B PTS: 1 7. ANS: C PTS: 1 8. ANS: D PTS: 1 SHORT ANSWER 9. ANS: Atomic Structure is the number of protons found in the nucleus.

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Chapter 4 (Atomic Structure) Test Study Guide An atom is the smallest particle of an element that retains the properties of that element. Democritus lived from 460

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Atomic Structure Study Guide Answers Atomic Structures Study Guide ANSWERS 1. The Periodic Table

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is mostly made of metals. 2. The noble gases are the farthest group to the right on the Periodic Table. 3. Atoms having equal numbers of positive and negative charges make them neutral. 4.

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Mass number ----- while atomic mass-----Mass number is the total number of protons and neutrons in a particular atom, while atomic mass is the average mass of all of the isotopes of a particular element. How do isotopes form? Isotopes are formed because every atom of an element do NOT have the same number of neutrons in the nucleus.

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